Additions and Corrections

1991, Volume 30

John H. Yamamoto, Wesley Yoshida, and Craig M. Jensen*: Unusual Reversible Dimerization of a μ-Pyridine-2-thiolato (pyS) Complex: Crystal Structure of $Pd_2(\mu\text{-}N\text{-}S\text{-}\eta^2\text{-pyS})_2Cl_2(PMe_3)_2$.

Pages 1353–1357. The energetics reported for the dimer–monomer interconversion involving $Pd_2(\mu\text{-}N\text{-}S\text{-}\eta^2\text{-pyS})_2Cl_2(PMe_3)_2$ were based on erroneous equilibrium constants. The correctly determined energetics are as follows: for the ^{31}P NMR study, ΔH = 52 kJ mol^{-1} and ΔS = 122 J K^{-1} mol^{-1}; for the ^1H NMR study, ΔH = 43 kJ mol^{-1} and ΔS = 107 J K^{-1} mol^{-1}.

1992, Volume 31

Glenn P. A. Yap and Craig M. Jensen*: Dynamic Behavior and Crystal Structures of $Pd_2(\mu_{NS},\eta^2\text{-L})_2Cl_2(PMe_3)_2$ (L = Pyrimidine-2-thiolate, 4-Methylpyrimidine-2-thiolate, and Methylimidazole-2-thiolate): Influence of the N–C–S Bond Angle on the Stability of (Heterocyclic 2-thiolato)dipalladium Complexes.

Pages 4823–4828. The energetics reported for the dimer–monomer interconversion involving $Pd_2(\mu_{NS},\eta^2\text{-primS})_2Cl_2(PMe_3)_2$ were based on erroneous equilibrium constants. The correctly determined energetics are ΔH = 42 kJ mol^{-1} and ΔS = 123 J K^{-1} mol^{-1}.

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